

Synthesis, Characterization, Structural and Phase Analysis of Novel Antiferroelectric Solid Solution



PRESENTER

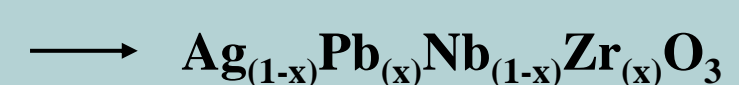
Yihan Lin, yla496@sfu.ca

Department of Chemistry, Faculty of Science

BACKGROUND

Antiferroelectric materials have drawn attentions widely, as promising candidates for energy storage and other applications. In this work, a novel antiferroelectric perovskite solid solution system, $(1-x)\text{AgNbO}_3$ - $(x)\text{PbZrO}_3$ will be introduced, along with its interesting phase transitions and relaxation behaviors.

METHOD



Hand grinding metal oxides PbO , ZrO_2 , Ag_2O and Nb_2O_5 raw powder

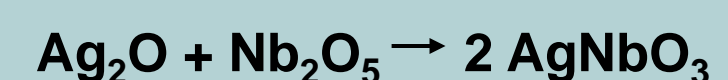
Press into large pellets

Calcine at 950°C for 6 hours (under O_2 atmosphere)

Break pellet and grind again

Add PVA, press into small pellets

Sinter at 1100°C for 10 hours (under O_2 atmosphere)



Place calcined AgNbO_3 powder into alumina crucible

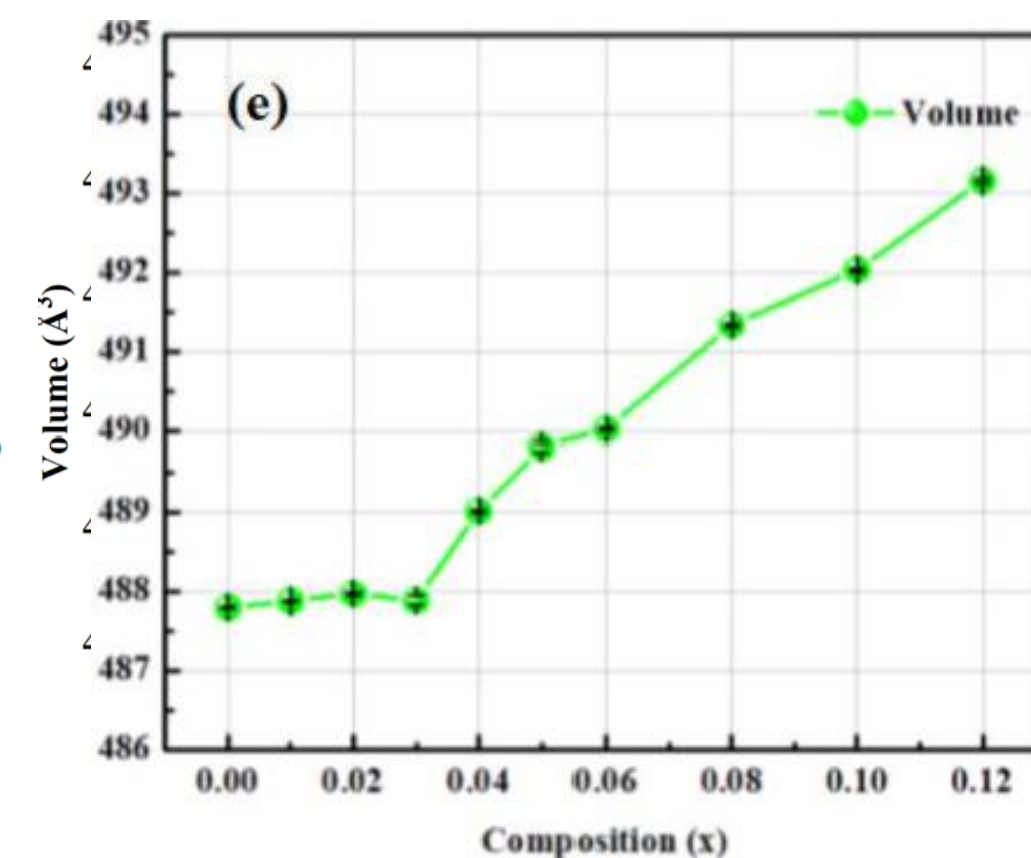
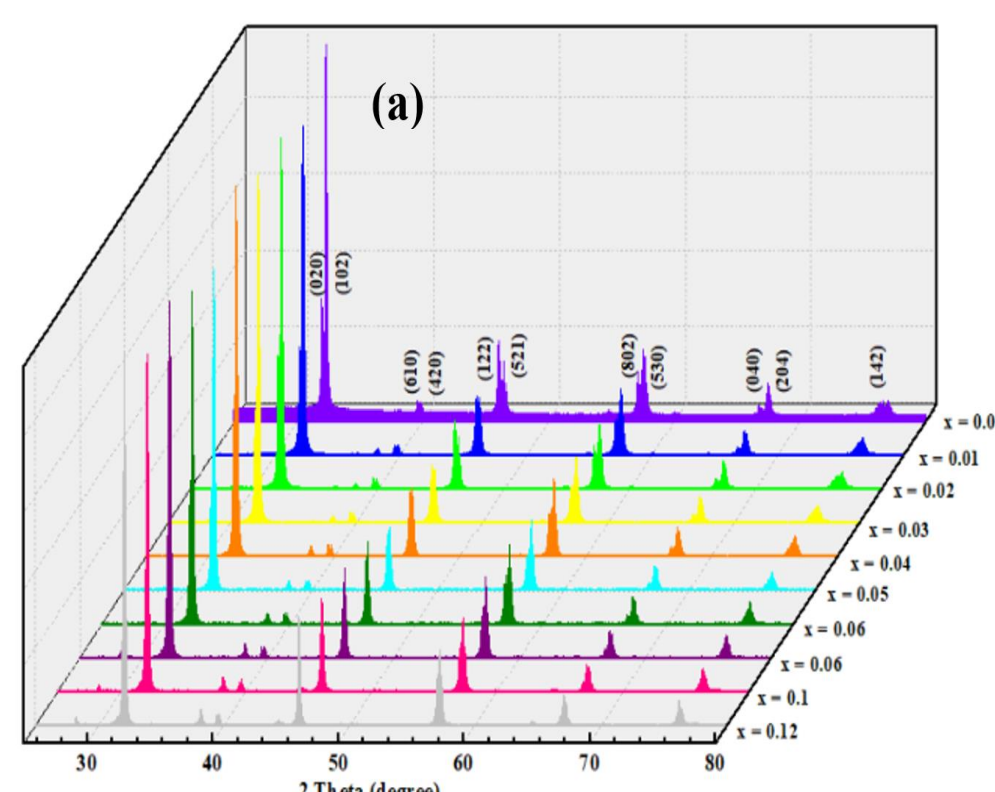
Heat furnace to 1300°C with rate 300°C/hr

Keep temperature at 1300°C for 4 hours

Cool furnace to room temperature at 20°C/hr

RESULTS & DISCUSSION

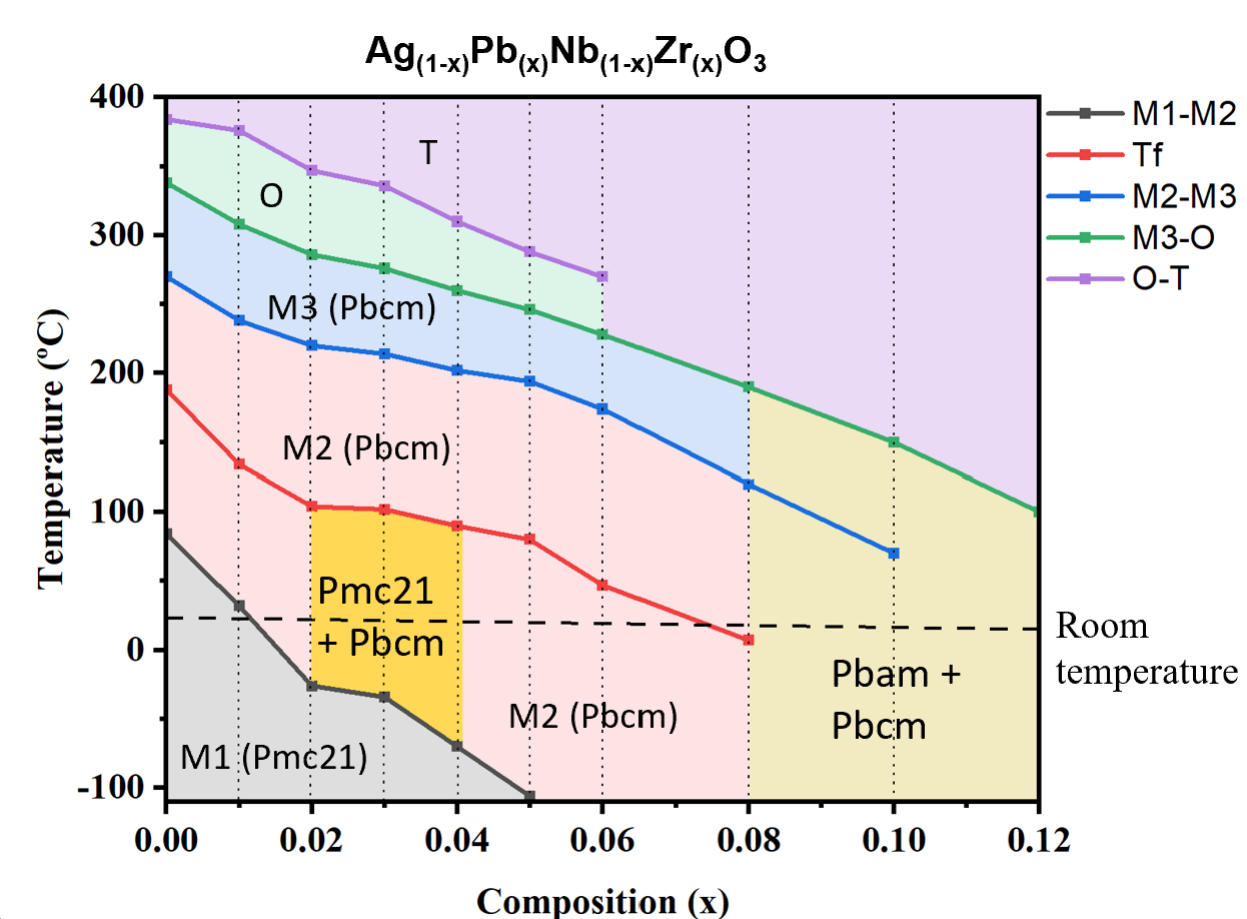
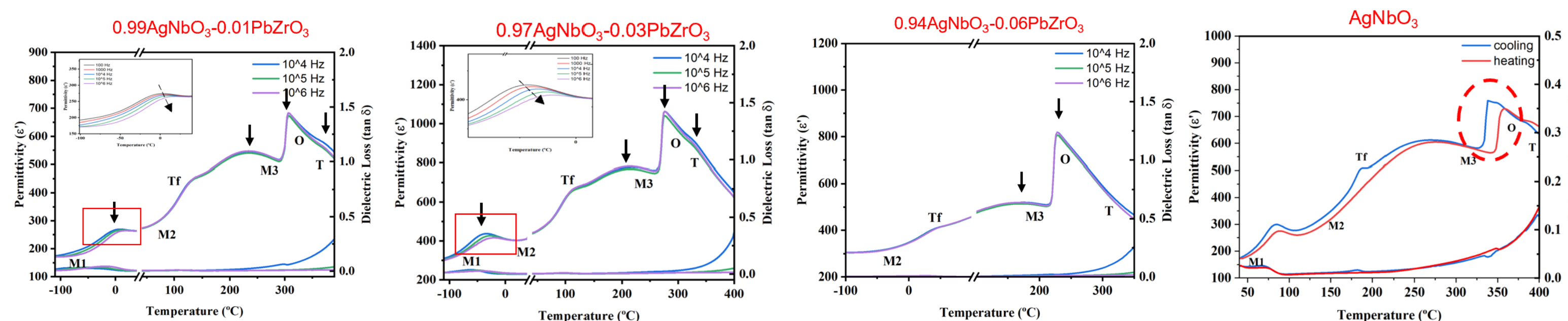
Structural Analysis



With the increase in concentration of lead zirconate PbZrO_3 :

- The main symmetry of the solid solution remains orthorhombic but having different space groups.
- Cell volume increased

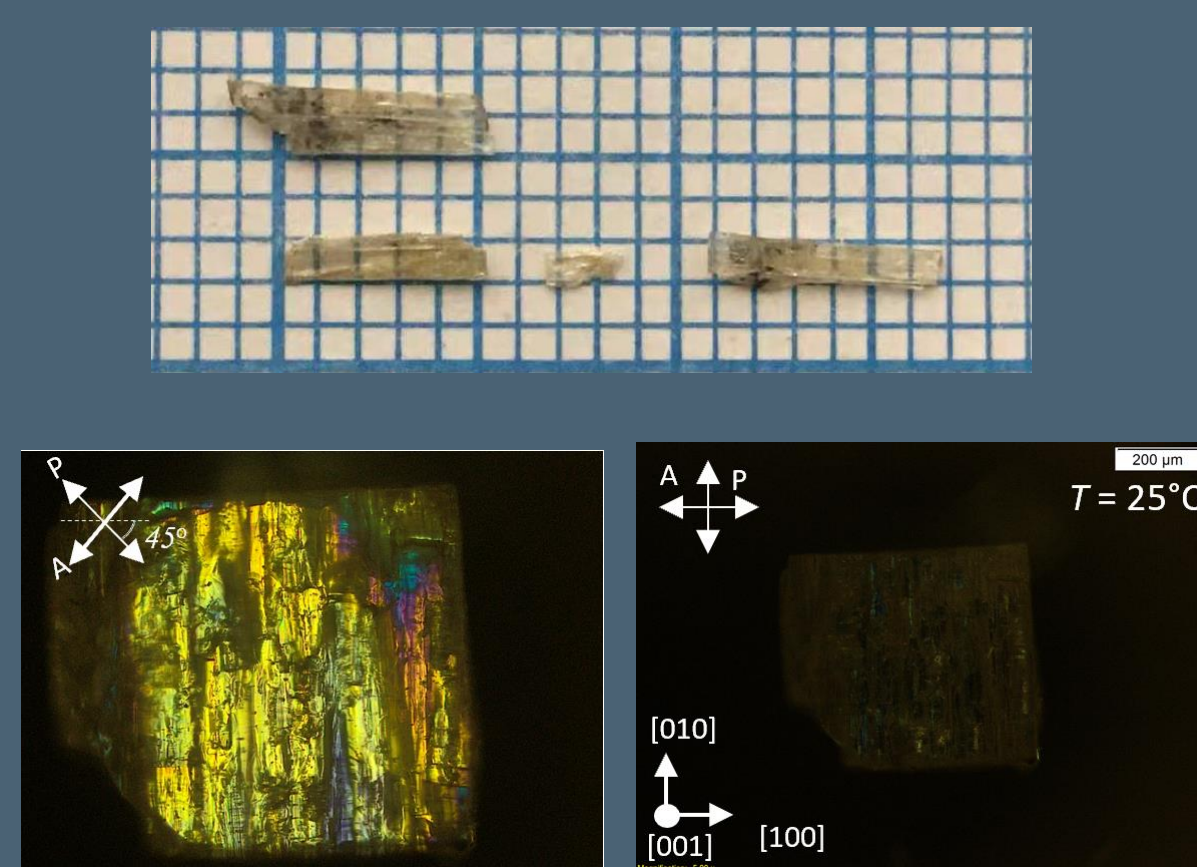
Dielectric Analysis



As the PbZrO_3 concentration increases

- All transition temperatures (as shown in arrows) shift lower
- Orthorhombic - Tetragonal (O-T) transition intensity diminished
- Dielectric thermal hysteresis behavior (circled in red) upon heating and cooling shown a first order phase transition

Single Crystals

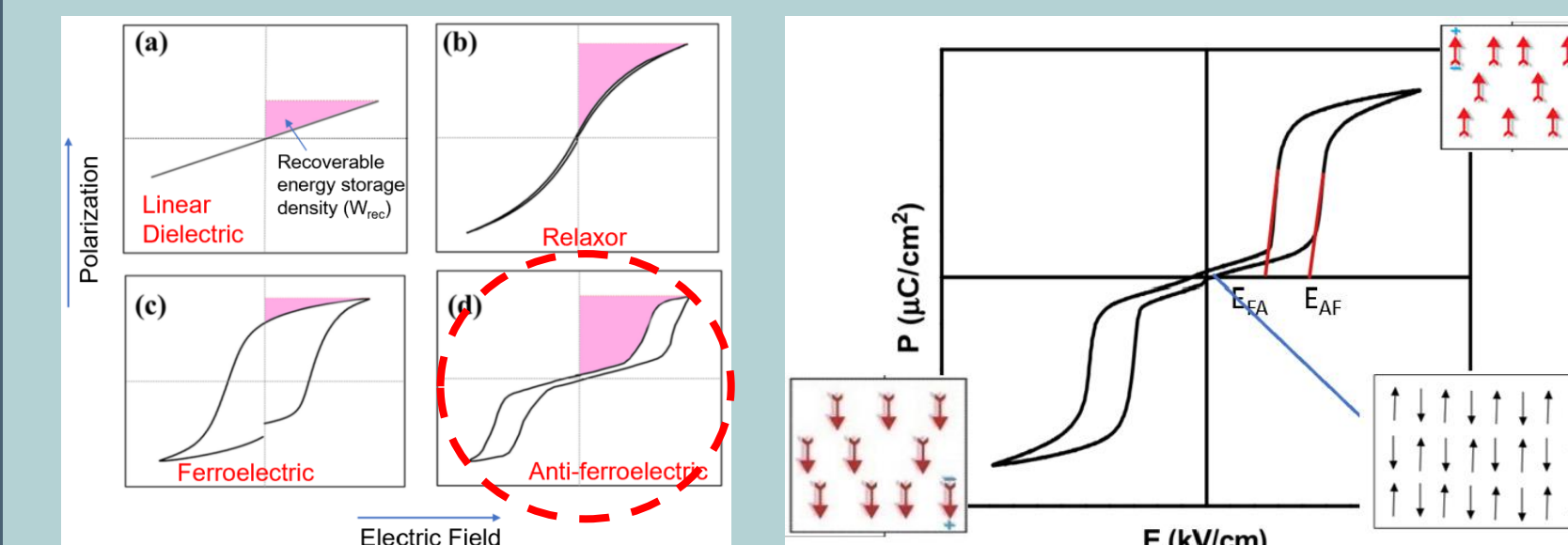
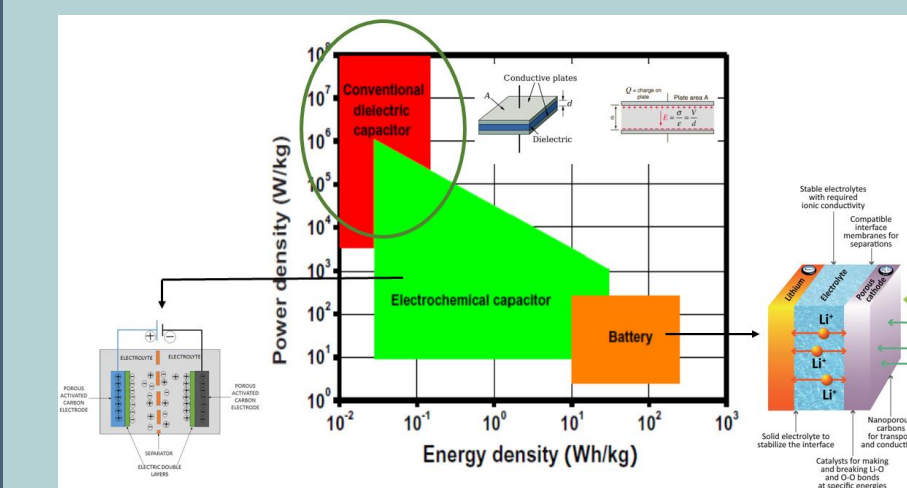


Extinction of light at 0° proved orthorhombic phase symmetry at room temperature

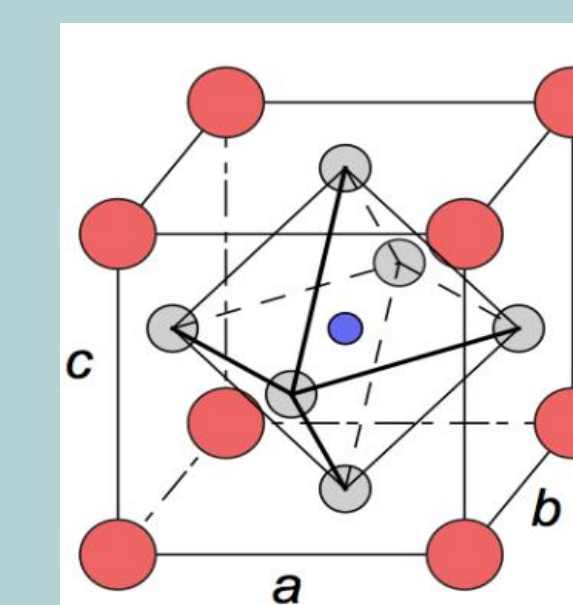
- As temperature increased, reflected color of crystal changed, indicating birefringence changes
- Moore study is needed to understand the optical properties

ADDITIONAL INFORMATION

Types of dielectric materials

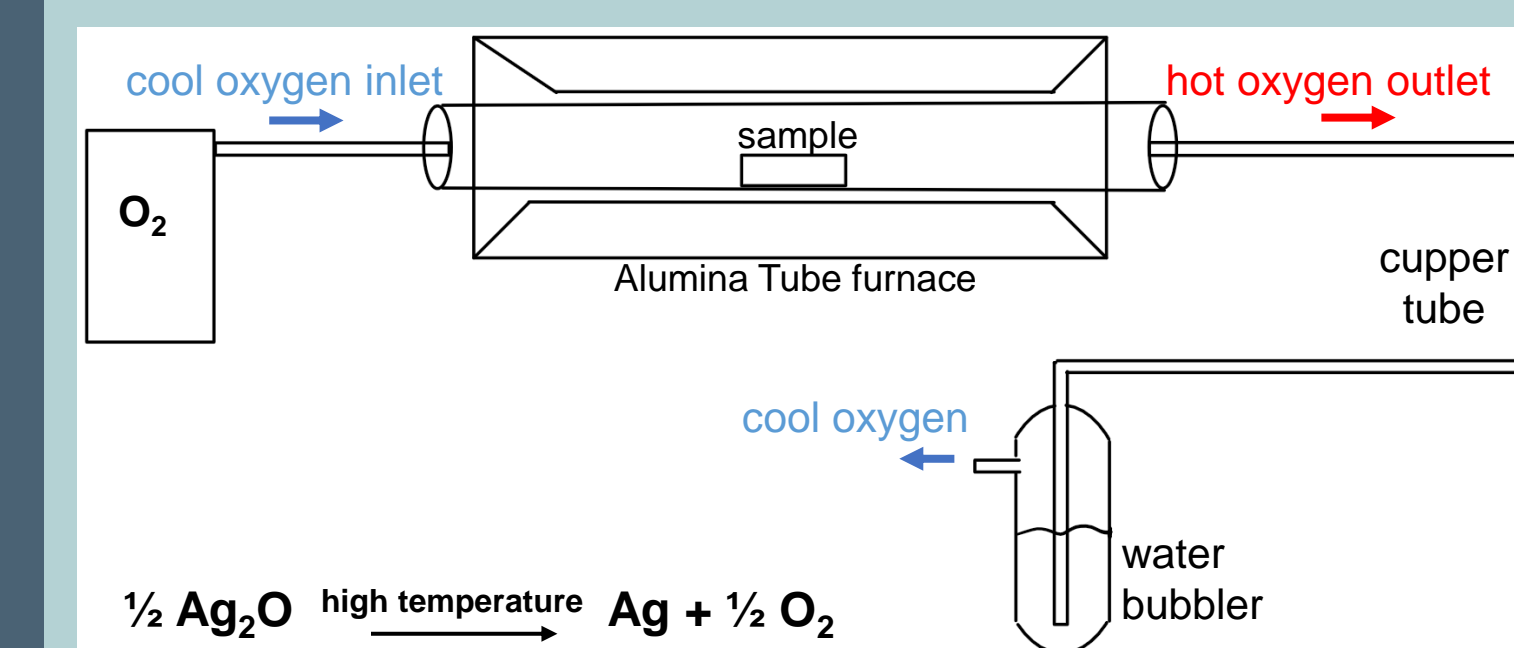


That is a perovskite structure



- A cations: large cations like Ag^+ , Pb^{2+} ...
- B cations: smaller cations like Nb^{5+} , Zr^{4+} ...
- Oxygen anion: O^{2-}

Experimental setup



REFERENCES

- J. Gao et al., J. Am. Ceram Soc. 101, 5443–5450 (2018)
- Z. Liu et al., Adv. Mater. Technol. 3, 1800111 (2018)
- P. Gao, Z.-G. Ye et al., Chem. Mater. 31, 979–990 (2019)
- H. J. Goldschmidt AND J. R. Rait, Nature, 152, 356–356, (1943).

ACKNOWLEDGEMENTS



undergraduate research symposium

